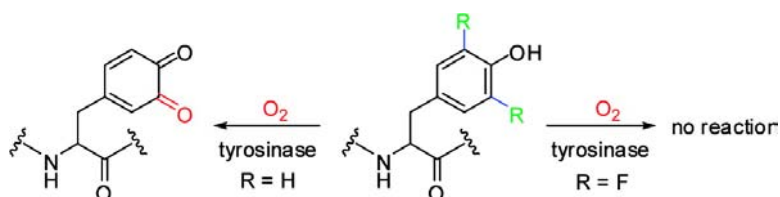


## Synthesis of 3,5-Difluorotyrosine-Containing Peptides: Application in Substrate Profiling of Protein Tyrosine Phosphatases

Bhaskar Gopishetty, Lige Ren, Tiffany M. Waller, Anne-Sophie Wavreille, Miguel Lopez, Amit Thakkar, Jinge Zhu, and Dehua Pei

*Org. Lett.*, **2008**, 10 (20), 4605-4608 • DOI: 10.1021/ol801868a • Publication Date (Web): 18 September 2008

Downloaded from <http://pubs.acs.org> on March 24, 2009



### More About This Article

Additional resources and features associated with this article are available within the HTML version:

- Supporting Information
- Access to high resolution figures
- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article

[View the Full Text HTML](#)

# Synthesis of 3,5-Difluorotyrosine-Containing Peptides: Application in Substrate Profiling of Protein Tyrosine Phosphatases

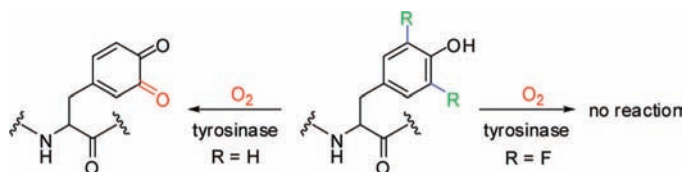
Bhaskar Gopishetty, Lige Ren, Tiffany M. Waller, Anne-Sophie Wavreille, Miguel Lopez, Amit Thakkar, Jinge Zhu, and Dehua Pei\*

Department of Chemistry and Ohio State Biochemistry Program, The Ohio State University, 100 West 18th Avenue, Columbus, Ohio 43210

pei.3@osu.edu

Received August 14, 2008

## ABSTRACT



Fully protected 3,5-difluorotyrosine (F<sub>2</sub>Y), Fmoc-F<sub>2</sub>Y(tBu)-OH, is efficiently prepared by a chemoenzymatic process and incorporated into individual peptides and combinatorial peptide libraries. The F<sub>2</sub>Y-containing peptides display kinetic properties toward protein tyrosine phosphatases (PTPs) similar to their corresponding tyrosine-containing counterparts but are resistant to tyrosinase action. These properties make F<sub>2</sub>Y a useful tyrosine surrogate during peptide library screening for optimal PTP substrates.

Replacement of a hydrogen with fluorine results in a small increase in the molecular size but often dramatically different physical, chemical, and biological properties of the molecule.<sup>1</sup> As such, fluorinated compounds provide useful mechanistic probes of enzyme-catalyzed reactions and other biological processes. For example, ring fluorinated analogues of tyrosine have been used to examine the catalytic mechanisms of tyrosinase,<sup>2</sup> tyrosine phenol-lyase,<sup>3</sup> protein tyrosine kinase and phosphatase,<sup>4,5</sup>  $\Delta 5$ -3-ketosteroid isomerase,<sup>6</sup> and ribonucleotide reductase.<sup>7</sup>

Our own interest in fluorotyrosines stems from our ongoing studies on protein tyrosine phosphatases (PTPs), a large

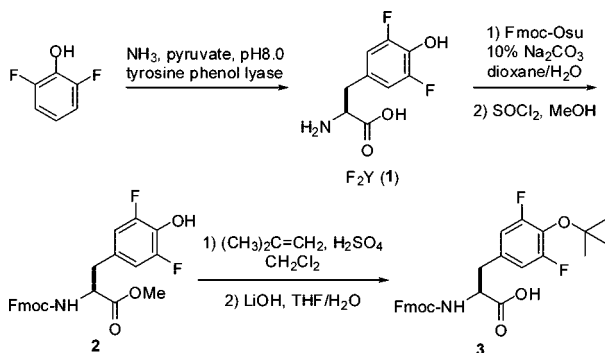
family of enzymes that catalyze the hydrolysis of phosphotyrosine (pY) in proteins back to tyrosine and inorganic phosphate. A major challenge in the PTP field is to determine the physiological substrates and cellular function of these enzymes. We recently developed a combinatorial library method to profile the substrate specificity of PTPs and subsequently used the specificity information to predict the protein substrates of the PTPs.<sup>8</sup> A key element of the technique involved selective derivatization of the reaction product (i.e., tyrosine) with a chemical tag (e.g., biotin). This was accomplished by first oxidizing the tyrosine side chain into an orthoquinone with O<sub>2</sub> and tyrosinase, followed by conjugate addition with biotin-hydrazide. To prevent false positives from unreacted substrates, tyrosine was excluded from the library. However, PTPs may require tyrosine residues for optimal binding and catalysis. Thus, we envisaged the inclusion of a fluorinated tyrosine into the peptide

(1) Hagmann, W. K. *J. Med. Chem.* **2008**, *51*, 4359–4369.  
(2) Battaini, G.; Monzani, E.; Casella, L.; Lonardi, E.; Tepper, A. W. J. W.; Canters, G. W.; Bubacco, L. *J. Biol. Chem.* **2002**, *277*, 44606–44612.  
(3) Phillips, R. S.; Von Tersch, R. L.; Secundo, F. *Eur. J. Biochem.* **1997**, *244*, 658–663.  
(4) Kim, K.; Cole, P. A. *J. Am. Chem. Soc.* **1998**, *120*, 6851–6858.  
(5) Martin, B.; Pallen, C. J.; Wang, J. H.; Graves, D. J. *J. Biol. Chem.* **1985**, *260*, 14932–14937.  
(6) Brooks, B.; Phillips, R. S.; Benisek, W. F. *Biochemistry* **1998**, *37*, 9738–9742.

(7) Yee, C. S.; Chang, M. C. Y.; Ge, J.; Nocera, D. G.; Stubbe, J. *J. Am. Chem. Soc.* **2003**, *125*, 10506–10507.  
(8) Garaud, M.; Pei, D. *J. Am. Chem. Soc.* **2007**, *129*, 5366–5367.

library as a tyrosinase-resistant tyrosine surrogate. Among all of the known mono-, di-, and multiply fluorinated tyrosine analogues, we reasoned that 3,5-difluorotyrosine (F<sub>2</sub>Y, compound **1** in Scheme 1) should contain the minimal

**Scheme 1.** Synthesis of Fully Protected F<sub>2</sub>Y



number of fluorine substitutions to render it resistant to tyrosinase action. Because F and H atoms have similar van der Waals radii (1.35 Å for F and 1.10 Å for H), Tyr and F<sub>2</sub>Y are essentially isosteric, although the side chain of F<sub>2</sub>Y has a lower pK<sub>a</sub> value (7.2) than that of Tyr (9.9).<sup>9</sup>

F<sub>2</sub>Y has previously been synthesized both chemically<sup>10</sup> and enzymatically.<sup>3,4,7</sup> Synthesis of F<sub>2</sub>Y-containing peptides employed side chain unprotected Fmoc-F<sub>2</sub>Y-OH, and the resulting peptides were purified by HPLC.<sup>4,7</sup> The reported peptides were either very short or contained F<sub>2</sub>Y near their N-termini (therefore no repeated coupling reactions after incorporation of F<sub>2</sub>Y), and each contained only a single F<sub>2</sub>Y residue. We felt that the unprotected F<sub>2</sub>Y side chain might be problematic with peptide library synthesis, during which more forcing coupling conditions are typically employed to drive reactions to completion, some library members will contain multiple F<sub>2</sub>Y residues, and HPLC purification is not an option. In this report, we describe the synthesis of fully protected F<sub>2</sub>Y, its incorporation into peptides and peptide libraries, and its activity against PTPs in comparison with the tyrosine counterparts.

We employed the enzymatic synthesis originally developed by Phillips and co-workers<sup>3</sup> to prepare multigram quantities of F<sub>2</sub>Y from 2,6-difluorophenol, pyruvate, and NH<sub>3</sub> (Scheme 1). Treatment of F<sub>2</sub>Y with *N*-(9-fluorenylmethylcarbonyloxy)succinimide in 10% sodium carbonate gave N<sup>α</sup>-Fmoc-F<sub>2</sub>Y-OH, which was subsequently converted into its methyl ester **2** using thionyl chloride in refluxing methanol. The phenol group was next protected as a *tert*-butyl ether by treatment with isobutylene and H<sub>2</sub>SO<sub>4</sub>. Finally, hydrolysis of the methyl ester by LiOH in THF/H<sub>2</sub>O gave the desired N<sup>α</sup>-Fmoc-F<sub>2</sub>Y(*t*Bu)-OH (**3**) in 24% overall yield (from F<sub>2</sub>Y).

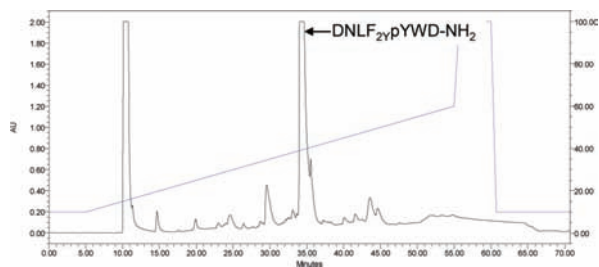
Three F<sub>2</sub>Y-containing peptides and their corresponding Tyr-containing counterparts were synthesized on the solid

phase by using standard Fmoc/HBTU chemistry (Table 1,

**Table 1.** Comparison of the Kinetic Constants of F<sub>2</sub>Y- and Tyr-Containing Peptides against PTP1B (pH 7.4)

peptide	$k_{\text{cat}}$ (s <sup>-1</sup> )	$K_M$ (μM)	$k_{\text{cat}}/K_M$ (μM <sup>-1</sup> s <sup>-1</sup> )
DNLF <sub>2Y</sub> pYWD-NH <sub>2</sub> ( <b>4</b> )	36 ± 1	3.6 ± 0.4	10.0
DNLYpYWD-NH <sub>2</sub> ( <b>5</b> )	32 ± 1	5.8 ± 0.8	5.5
DDTF <sub>2Y</sub> DpYAA-NH <sub>2</sub> ( <b>6</b> )	36 ± 2	13 ± 3	2.7
DDTYDpYAA-NH <sub>2</sub> ( <b>7</b> )	33 ± 1	14 ± 2	2.3
REF <sub>2Y</sub> EFpYAA-NH <sub>2</sub> ( <b>8</b> )	44 ± 2	14 ± 2	3.2
REYEFpYAA-NH <sub>2</sub> ( <b>9</b> )	33 ± 1	6.0 ± 0.7	5.4

compounds **4–9**). Peptide **5** is derived from a known pY motif of receptor protein tyrosine kinase erbB2,<sup>11</sup> whereas peptides **7** and **9** are analogous to the preferred substrates of PTP1B (the prototypical PTP), previously identified from a peptide library.<sup>8</sup> These peptides contain F<sub>2</sub>Y (or Tyr) at the pY-3, pY-2, or pY-1 position (relative to pY, which is designated as position 0). Earlier studies have shown that residues at the N-terminal side of pY are most critical in defining the substrate specificity of PTPs.<sup>12</sup> During peptide synthesis, N<sup>α</sup>-Fmoc-F<sub>2</sub>Y(*t*Bu)-OH was efficiently incorporated into peptides without incidence (as judged by ninhydrin tests), despite that it was used at only 1.5 equiv (4 equiv was used for all other amino acids). Reversed-phase HPLC analysis of the crude peptides showed that in each case the desired peptide was the major product (see Figure 1, for



**Figure 1.** HPLC analysis of peptide **4** on a C-18 column, eluted with a linear gradient of 10–60% CH<sub>3</sub>CN in H<sub>2</sub>O containing 0.1% trifluoroacetic acid over 50 min.

example). Peptides **4–9** were purified by preparative HPLC, and their kinetic activities toward PTP1B (i.e.,  $k_{\text{cat}}$ ,  $K_M$ , and  $k_{\text{cat}}/K_M$  values) were determined at pH 7.4 (Table 1). The data show that substitution of F<sub>2</sub>Y for tyrosine resulted in minimal changes in kinetic constants (≤2-fold, which is within the margin of experimental error). Thus, F<sub>2</sub>Y is a good functional mimic of tyrosine, in terms of binding to the active site of PTPs.

(11) Yamamoto, T.; Ikawa, S.; Akiyama, T.; Nomura, N.; Miyajima, N.; Saito, T.; Toyoshima, K. *Nature* **1986**, *319*, 230–234.

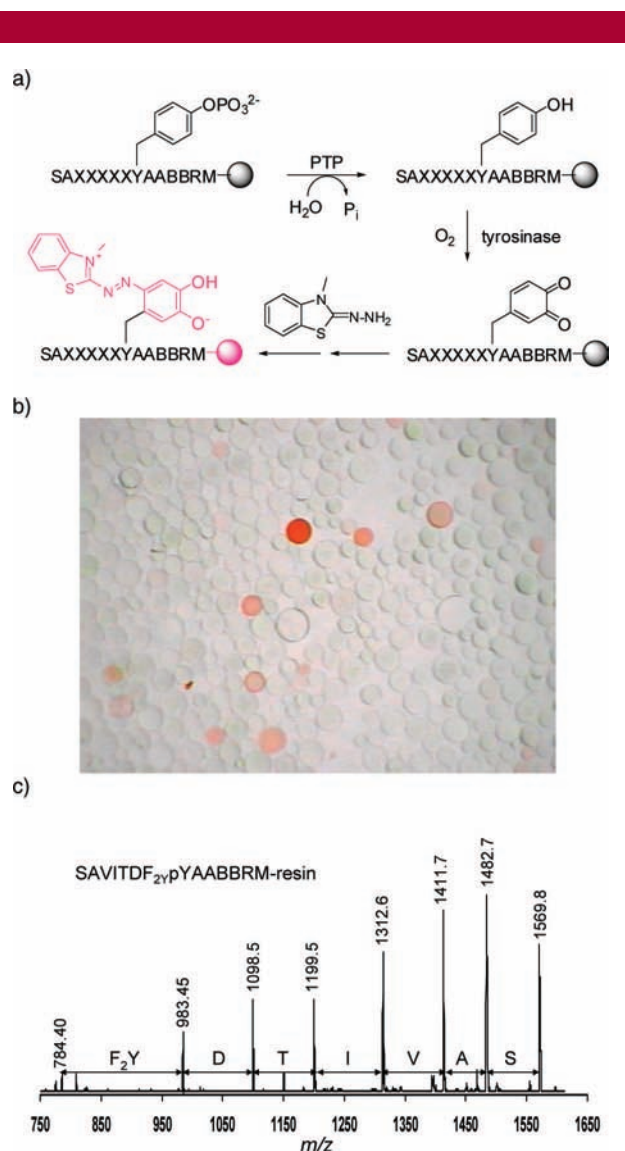
(12) (a) Zhang, Z.-Y.; Thieme-Sefler, A. M.; Maclean, D.; McNamara, D. J.; Dobrusin, E. M.; Sawyer, T. M.; Dixon, J. E. *Proc. Natl. Acad. Sci. U.S.A.* **1993**, *90*, 4446–4450. (b) Zhang, Z.-Y.; Maclean, D.; McNamara, D. J.; Sawyer, T. K.; Dixon, J. E. *Biochemistry* **1994**, *33*, 2285–2290.

(9) Seyedsayamdost, M. R.; Reece, S. Y.; Nocera, D. G.; Stubbe, J. *J. Am. Chem. Soc.* **2006**, *128*, 1569–1579.

(10) Kirk, K. L. *J. Org. Chem.* **1980**, *45*, 2015–2016.

Next, a pair of Tyr- and F<sub>2</sub>Y-containing peptides (peptides **8** and **9** in Table 1) were tested for activity against tyrosinase. The peptides were individually treated with tyrosinase in the presence of atmospheric O<sub>2</sub> and excess biotin hydrazide. Reversed-phase HPLC/MS analysis of the reaction mixtures revealed that the Tyr-containing peptide REYEFpYAA was quantitatively converted into a new species, which had an increased retention time (from 32.0 to 34.5 min) and molecular mass (from *m/z* 1168.4. to 1440.7) (Figure S1 in Supporting Information). The increase of 272.3 amu in molecular mass is consistent with the addition of a single biotin hydrazide molecule to the tyrosine side chain. As expected, the F<sub>2</sub>Y-containing peptide (**8**) was completely resistant to tyrosinase action.

Finally, to test whether F<sub>2</sub>Y is compatible with peptide library synthesis, screening, and postscreening sequence identification, we designed a pY peptide library containing five random positions immediately N-terminal to pY [SAXXXXXpYAABBRM-resin, where B is β-alanine and X is F<sub>2</sub>Y, norleucine (Nle), or any of the 17 proteinogenic amino acids excluding Met, Cys, and Tyr]. The library was synthesized on polyethylene glycol polyacrylamide (PEGA) resin<sup>13</sup> by the split-and-pool method.<sup>14</sup> A portion of the resulting one-bead-one-compound (OBOC) library (~40 000 beads) was treated with 1.0 nM PTP1B for 20 min (at pH 7.4), followed by incubation with 1.2 μM mushroom tyrosinase in the presence of atmospheric O<sub>2</sub> and 6 mM 3-methyl-2-benzothiazolinonehydrazide (MBTH). Under these conditions, only beads that carried the most preferred substrates of PTP1B were dephosphorylated (usually only partially). The exposed tyrosine side chain was subsequently oxidized by the excess tyrosinase activity into an ortho-quinone, which was trapped by reaction with MBTH to form a dark pink pigment (Figure 2a).<sup>15</sup> Thus, as a result of this reaction cascade, a bead carrying a preferred PTP1B substrate would turn pink/red, whereas beads carrying poor substrates would not. This was indeed the case (Figure 2b). The pink/red colored beads were removed from the library and individually sequenced by partial Edman degradation/mass spectrometry<sup>16</sup> to reveal the most preferred PTP1B substrates (Table 2). Figure 2c shows an example of the MS spectrum derived from one of the pink beads (bead No. 2 in Table 2), carrying the sequence of SAVITDF<sub>2</sub>YpYAABBRM. Our results are in agreement with earlier reports that PTP1B prefers acidic residues at the N-terminal side of pY, especially at the pY-2 position.<sup>8,12,17</sup> A control



**Figure 2.** (a) Reactions involved in library screening against PTP and bead coloration. (b) A portion of the OBOC pY library after treatment with PTP1B (1.0 nM) and tyrosinase (1.2 μM) (viewed under a dissecting microscope). (c) MALDI-TOF spectrum of peptide SAVITDF<sub>2</sub>YpYAABBRM\* and its truncation products after partial Edman degradation (derived from a single PEGA resin bead). M\*, homoserine lactone.

**Table 2.** Partial List of the Selected PTP1B Substrates<sup>a</sup>

bead no.	peptide sequence
1	QDVDApYAA
2	VITDF <sub>2</sub> YpYAA
3	LQF <sub>2</sub> YpYAA
4	ITMDQpYAA
5	WGTDSpYAA
6	SSFVpYAA
7	SRHEWpYAA
8	ETDFApYAA
9	NDLF <sub>2</sub> YpYAA
10	FTSGLpYAA

<sup>a</sup> M, norleucine.

(13) Meldal, M. *Tetrahedron Lett.* **1992**, 33, 3077–3080.  
 (14) (a) Lam, K. S.; Salmon, S. E.; Hersh, E. M.; Hruby, V. J.; Kazmierski, W. M.; Knapp, R. J. *Nature* **1991**, 354, 82–84. (b) Houghten, R. A.; Pinilla, C.; Blondelle, S. E.; Appel, J. R.; Dooley, C. T.; Cuervo, J. H. *Nature* **1991**, 354, 84–86. (c) Furka, A.; Sebestyen, F.; Asgedom, M.; Dibo, G. *Int. J. Pept. Protein Res.* **1991**, 37, 487–493.  
 (15) Winder, A. J.; Harris, H. *Eur. J. Biochem.* **1991**, 198, 317–326.  
 (16) Thakkar, A.; Wavreille, A.-S.; Pei, D. *Anal. Chem.* **2006**, 78, 5935–5939.  
 (17) (a) Pellegrini, M. C.; Liang, H.; Mandiyan, S.; Wang, K.; Yuryev, A.; Vlattas, I.; Sytwu, T.; Li, Y.-C.; Wennogle, L. P. *Biochemistry* **1998**, 37, 15598–15606. (b) Vetter, S. W.; Keng, Y.-F.; Lawrence, D. S.; Zhang, Z.-Y. *J. Biol. Chem.* **2000**, 275, 2265–2268. (c) Walchli, S.; Espanel, X.; Harrenga, A.; Rossi, M.; Cesareni, G.; van Huijsduinen, R. H. *J. Biol. Chem.* **2004**, 279, 311–318.

experiment without the PTP1B treatment resulted in no pink/red colored beads at all (data not shown). Since ~25% of the library members contain at least one F<sub>2</sub>Y residue in the random region, the lack of any pink/red beads in the control provides further evidence that F<sub>2</sub>Y is completely resistant to tyrosinase action. Most importantly, F<sub>2</sub>Y is compatible with every aspect of library synthesis, screening, and sequence identification.

In summary, we have found that F<sub>2</sub>Y is a functional, tyrosinase-resistant mimetic of tyrosine for applications such as defining the substrate specificity of PTPs through combinatorial library screening. An efficient method has been developed for the chemoenzymatic synthesis of fully protected F<sub>2</sub>Y and its incorporation into peptides and peptide libraries. In addition, an improved PTP screening assay has

been developed by employing MBTH as the coupling reagent. Because the pink/red pigment is only formed on the beads that have undergone PTP reaction, the current method does not generate any false positives and is operationally straightforward. Application of the methodologies to systematically profile the substrate specificity of PTPs is currently ongoing in our laboratory and will be reported elsewhere in due course.

**Acknowledgment.** This work was supported by the National Institutes of Health (GM062820).

**Supporting Information Available:** Detailed experimental procedure and spectral data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL801868A